

Sample Exam 1 Name Chem 212

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Sample Exam 1 Name Chem

Exam 1 Name CHEM 210 - Pennsylvania State University

Exam 1 Name ____ CHEM 210 1 (6 pts) Give each non-hydrogen atom the proper formal charge, even if zero; all electrons are shown a b c d 2 (9 pts)
Draw one

Sample Exam 1 Name CHEM 212 - Pennsylvania State University

Sample Exam 1 Name ____ CHEM 212 1 (20 pts) Synthesis Show (using one or more steps) how the following compounds can be synthesized from the

CHEM 150 Exam 1 KEY Name Multiple Choice Identify the ...

CHEM 150 Exam 1 ____KEY____ Name Multiple Choice Identify the letter of the choice that best completes the statement or answers the question
B 1 Which of the following is synonymous with "fact"? a a hypothesis b an observation which is reproducible c an observation which is not reproducible d none of these _D_ 2

CHEM 293 Sample Exam 1 Name Dr. H.M. Muchall

CHEM 293 1 CHEM 293 ____ Sample Exam 1 Name Dr HM Muchall Remember: about a point a minute! This is about the length you can expect 1 It was proposed to use UV-Vis spectroscopy to follow the progress of the photochemical reaction O O

Chem. 111 Name EXAM 1 (Last) (First) - cerritos.edu

1 Chem 111 Name ____ Spring 07 EXAM 1 (Last) (First) 100 Points All significant figures and set-ups must be shown in an organized way All units must be shown throughout the problems Use the following table for molar masses Compound Molar Mass g/mole Compound Molar Mass g/mole H ...

Chem 260/261 First Exam 2/12/01 Sample

Name Chem 260/261 First Exam 2/12/01 Sample 2 Multiple Choice (42 pts): 14 questions Choose the letter corresponding to the best single answer

for each question Put your answers in the numbered boxes on the cover sheet of this exam SAMPLE QUESTIONS: 1)

Part I. Circle your answers - Colby College

CH141, Fall 2016 Practice Exam 1 Name: _____ Part I Circle your answers 1 If a sample of matter is uniform throughout and cannot be separated into other substances by physical means, it is _____ A) a compound B) either a compound or an element C) a homogeneous mixture D) a ...

CHEM 2481 Exam I Name - onu.edu

CHEM 2481 EXAM I SPRING 2018 3 10 Use a Maclaurin series to show that $\sin q \approx q$ for small values of q 11 Show that the wave function $y(x) = C \sin(kx)$, where C and k are constants, is a solution to the wave equation 12 Aerogels are synthetic solid materials having ...

Houston Community College System Chemistry 1411

EXAM # 3A Sample 2 CHEM 1411 EXAM # 3 Name: _____ (Chapters 8, 9,10,and 11) Score: Directions- please answer the following multiple-choice questions next to each number Please show all your work for bonus question and part(2) questions in the space provided Part (1) - Multiple Choice - (3 points each)

CHEM 1411 Exam #2 - Houston Community College

CHEM 1411 Exam # 2A Sample Name: _____ (Chapters 4,5,6, and 7) Score: Directions- please answer the following multiple-choice questions next to each number Please show all your work for bonus question and part (2) questions in the space provided

Chem 116 Sample Examination #1

Chem 116 Sample Examination #1 This exam consists of seven (7) pages, including this cover page Be sure your copy is complete before beginning your work If this test packet is defective, ask for another one A copy of the Periodic Table is attached at the back of the exam You may remove it and use the

Name Exam No. W CHEM 60 Fall 2016 Exam 1 Ch 1-4, 100 ...

Name _____ Exam No _____ W CHEM 60 Fall 2016 Exam 1 Ch 1-4, 100 points total Multiple Choice (20 questions, 3 points each = 60 points total) Mark the letter on the scantron form corresponding to the one best answer for each question or problem Write

Chem 115 Sample Examination #1

Chem 115 Sample Examination #1 This exam consists of seven (7) pages, including this cover page Be sure your copy is complete before beginning your work If this test packet is defective, ask for another one A copy of the Periodic Table is attached to the end of this exam You may remove it and use the back side as scratch paper

Chem 454 - Exam 1 - 2008 - University of Idaho

1 Chem 454 - instrumental Analysis - Exam 1 - February 6th, 2008 Name: _____ 1] A glass electrode was immersed into a solution of pH 4.33 gave a response of 6771 mV This electrode was used to measure a sample solution and gave a response of 5447 mV

Chem 128, Exam I Name - University of Connecticut

Chem 128, Exam I Name February 22, 2001 I (40 points) A Consider the following beakers: - Beaker A has 100 kg of a solvent Y and 0.100 mol of glucose (C₆H₁₂O₆) - Beaker B has 100 kg of a solvent Y and 0.100 mol of CaCl₂ - Beaker C has 100 kg of water and 0.100 mol of sucrose (C₁₂H₂₂O₁₁) Answer the questions below, using LT (for is less than), GT (for is greater than), EQ